

Illinois Institute of Technology
Physics

M.Sc. Comprehensive and Ph.D. Qualifying Examination

PART I

Thursday, August 22, 2019

4:00{8:00 PM

General Instructions

1. Each problem is to be done on a separate booklet. Label the front of each book with the identifying code letter you picked, the part number of the exam, and the number of the problem only; for example: A-I.6. Do not write your name or IIT ID number on any material handed in for grading.
2. Any numerical data not speci ed in a problem should be found in the table of constants at the front of the exam.
3. *DON'T PANIC*: It is not expected that each student will completely solve every problem. However, it is advisable to do a thorough job on those problems that you do solve.

Physical Constants

Speed of light in vacuum	$c = 2.998 \cdot 10^8 \text{ m/s}$
Planck's constant	$h = 6.626 \cdot 10^{-34} \text{ J s}$
	$\hbar = \frac{h}{2\pi}$
	$= 1.055 \cdot 10^{-34} \text{ J s}$
	$= 6.582 \cdot 10^{-16} \text{ eV s}$
Permeability constant	$\mu_0 = 4 \cdot 10^{-7} \text{ N/A}^2$
Permittivity constant	$\frac{1}{4\pi\epsilon_0} = 8.988 \cdot 10^9 \text{ N m}^2/\text{C}^2$
Fine structure constant	$= \frac{e^2}{4\pi\epsilon_0\hbar c}$
	$= 7.30 \cdot 10^{-3} = \frac{1}{137}$
Gravitational constant	$G = 6.67 \cdot 10^{-11} \text{ m}^3/\text{s}^2 \text{ kg}$
Avogadro's number	$N_A = 6.023 \cdot 10^{23} \text{ mole}^{-1}$
Boltzmann's constant	$k = 1.381 \cdot 10^{-23} \text{ J/K}$
	$= 8.617 \cdot 10^{-5} \text{ eV/K}$
kT at room temperature	$k \cdot 300 \text{ K} = 0.0258 \text{ eV}$
Universal gas constant	$R = 8.314 \text{ J/mole K}$
Stefan-Boltzmann constant	$= 5.67 \cdot 10^{-8} \text{ W/m}^2 \text{ K}^4$
Electron charge magnitude	$e = 1.602 \cdot 10^{-19} \text{ C}$
Electron rest mass	$m_e = 9.109 \cdot 10^{-31} \text{ kg}$
	$= 0.5110 \text{ MeV}/c^2$
Neutron rest mass	$m_n = 1.675 \cdot 10^{-27} \text{ kg}$
	$= 939.6 \text{ MeV}/c^2$
Proton rest mass	$m_p = 1.672 \cdot 10^{-27} \text{ kg}$
	$= 938.3 \text{ MeV}/c^2$
Deuteron rest mass	$m_d = 3.343 \cdot 10^{-27} \text{ kg}$
	$= 1875.6 \text{ MeV}/c^2$
Atomic mass unit ($C^{12} = 12$)	$u = 1.661 \cdot 10^{-27} \text{ kg}$
	$= 931.5 \text{ MeV}/c^2$
Mass of earth	$M_E = 5.98 \cdot 10^{24} \text{ kg}$
Radius of earth	$R_E = 6.37 \cdot 10^6 \text{ m}$
Mass of sun	$M_S = 1.99 \cdot 10^{30} \text{ kg}$
Radius of sun	$R_S = 6.96 \cdot 10^8 \text{ m}$
Gravitational acceleration at earth's surface	$g = 9.81 \text{ m/s}^2$
Atmospheric pressure	$= 1.01 \cdot 10^5 \text{ N/m}^2$
Radius of earth's orbit	$= 1.50 \cdot 10^{11} \text{ m}$
Radius of moon's orbit	$= 3.84 \cdot 10^8 \text{ m}$

Conversion Factors

1 eV	=	$1.602 \cdot 10^{-19} \text{ J}$	1 J	=	$6.242 \cdot 10^{18} \text{ eV}$
1 Å	=	10^{-10} m	1 Fermi	=	10^{-15} m
1 barn (b)	=	10^{-28} m^2	1 in	=	2.54 cm
0 °Celsius	=	273.16 K	1 cal	=	4.19 J

Problem 4: A satellite travels in a circular orbit of radius r_0 . Its rocket motor fires, suddenly increasing its velocity by 8% along its direction of motion. What is the apogee of the new orbit? Make a sketch superimposing the new orbit on the original orbit.

Problem 5: Consider a three-dimensional ideal gas placed into a spherically symmetric potential, given by the formula:

$$V(r) = \begin{cases} \frac{1}{2} U_0 \left(\frac{r}{R} \right)^2; & r \leq R \\ U_0 \ln(r/R); & r > R \end{cases}$$

(a) Find a single particle partition function.

A useful Gaussian integral:

$$\int_0^{\infty} x^2 e^{-x^2/a} dx = \frac{\sqrt{\pi}}{2} a^{3/2}$$

(b) Find an N -particle partition function for $N \gg 1$. A useful Stirling formula:

$$N! \approx \left(\frac{N}{e} \right)^N \sqrt{2\pi N}$$

in this case.

- (c) What is the highest possible gas temperature?
- (d) Find a (Helmholtz) free energy of the gas.
- (e) Find an internal energy of the gas.
- (f) Find a specific heat (capacity) of the gas.

Problem 6: Calculate a chemical potential for an ideal **two-dimensional** non-relativistic Fermi-gas of a surface density n_S and arbitrary temperature T . A fermion mass is equal to m and spin is $1/2$. Then find the chemical potential for a high and low temperature limits.

Problem 7: Using statistical mechanical principles, estimate the mean thickness of the earth's atmosphere in terms of the mass m of a nitrogen molecule, the gravitational acceleration g , and the average atmospheric temperature T . Evaluate your result numerically.

Problem 8:

- (a) A cylindrical glass optical fiber (radius R) with index of refraction, n , propagates light by total internal reflection at the air-glass interface. Give an expression for the maximum angle of incidence, θ , at the flat end of the fiber.